







HYDROGEN GENERATION

Our Products



SEPURAN° Noble	4" Cartridge	6" Cartridge	8" Cartridge
Stainless steel housings	SS316	SS316	SS316
Trans membrane pressure	40 bar / 580 psi	25 bar / 362 psi	80 bar / 1160 psi • 70 bar/1015 psi
Temperature	< 70 °C / 158 °F	< 70 °C / 158 °F	< 50 °C / 122 °F • < 70 °C / 158 °F

Recovery advantages

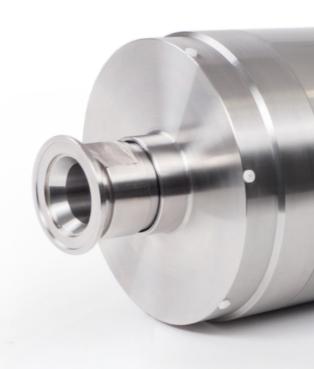
 Hydrogen revovery of more than 90 percent possible

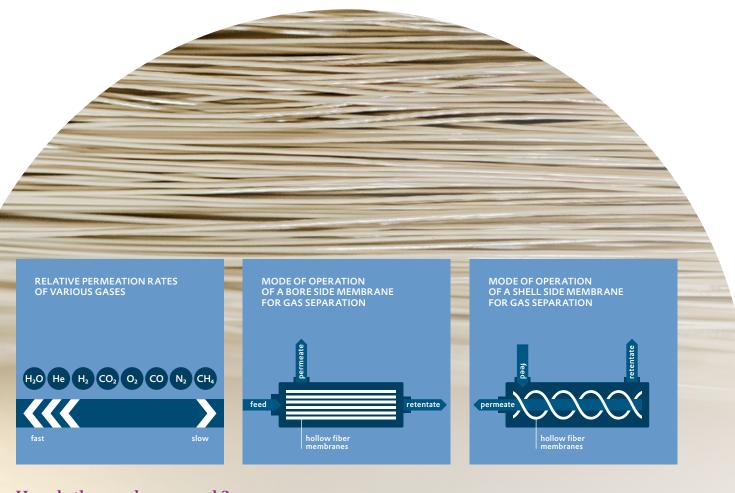
Upgrading advantages

- · High selectivity
- · Low energy consumption

Overall features

- Low space requirements
- Continuous separation process
- Simple modular setup
- · Flexible and easily expanded
- No other auxiliary materials, such as water and sorbents, required
- · No emissions into the environment





How do the membranes work?

Gas separation membranes work on the principle of selective permeation through a membrane surface. The driving force for permeation of the gas through the membrane is the difference between the partial pressures of the gas on the retentate side and the permeate side.

The greater this difference, the higher the proportion of the gas that permeates through the membrane. In a separation, such as between hydrogen and nitrogen, permeation of carbon dioxide through the membrane is much faster while nitrogen is retained within. The driving force required for the separation is obtained through a partial pressure gradient.

The permeation rate of each gas depends on its solubility in the membrane material and on the diffusion rate.

Gases that have higher solubility and smaller molecular size permeate the membrane faster than larger, less soluble gases. The ratio of the transport speeds of two gases is called selectivity.

The higher the selectivity, the higher the energy efficiency of the resulting membrane process.

Different membrane materials have different separation properties.



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• = registered trademark

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